# THE INELASTIC ROTOR SPECTROMETER AT THE HARWELL LINAC

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### INSTRUMENT DESCRIPTION

The Harwell Linac has been operating routinely for approximately 6 months at 25 Kw power and 75 Hz into a tantalum target. During this period a number of test experiments has been carried out and a period set aside for university user experiments.

The instrument viewed a water moderator in slab geometry at 25° to the normal. The moderator was 4.4 cm thick with gadolinium poisoning 13 mm below the surface. No decoupler was present. The measured coupling efficiency is  $5 \times 10^{-4} \, \text{n(1 eV)/n}_{\text{f/ster}}$ .

The spectrometer is designed to measure energy transfers from 50 meV to 400 meV covering a range of Q values from  $1-15 \text{ A}^{-1}$ . Particular emphasis has been placed on the low Q counter banks where measurements at low Q reduce multiphonon contributions in vibrational spectra, diffusional broadening in liquids and allow measurements to be made on magnetic excitations(1). The spectrometer consists of a Nimonic rotor, rotating at 600 Hz, accurately phased to the linac with a jitter of less than +  $\frac{1}{2} \mu s$ , placed at 6.4 metres from the moderator. Incident energy  $(E_{_{\scriptsize{O}}})$  selection is made by varying the phase of the rotor with respect to the linac.  $E_{\rm o}$  can be varied from 150 meV to 500 meV. The selected neutrons are allowed to fall on the sample placed 1 metre downstream in an evacuated chamber. Sample temperature can be controlled from 20K to room temperature. Low angle counter banks are placed between 40-110 either side of the main beam at 2.5 metres. In addition, counter banks are placed every 100 between 240-940 at 1.62 metres. counters are 1" diameter He3 10 atmosphere and 4 atmosphere respectively. The region of  $(Q,\omega)$  space covered for two values of  $E_Q$  is shown in Figure 1 together with that covered by a beryllium filter spectrometer operating at these energy transfers. The energy resolution is estimated to be 7 meV at an energy transfer of 150 meV (4.7%). The intensity at the sample was measured using vanadium scattering to be 1100 n/s over 2" x 1" beam at 450 meV E.

#### 2. EXPERIMENTAL RESULTS

Figure 2 shows the measured spectra for a sample of sodium bifluoride NaHF2. The sample was  $^{\circ}$  13% scatterer 3 mm thick inclined at 450 to the beam and kept at 90K. NaHF<sub>2</sub> in the pure salt form has been investigated before with neutrons but only at high Q values and relatively poor energy resolution using the beryllium filter technique (2). The bifluoride ion has a sharp bending mode  $v_2$  at 156 meV and an antisymmetric stretch  $v_3$  at  $^{\circ}$  177 meV. The two modes are separated in the pure salt by IR measurement (3) but are both broad due to interactions with near ions. In a dilute sample of HF2 ions in KCl the modes are seen as extremely narrow peaks in the IR spectrum. The modes have not been separated before in a neutron scattering experiment from the pure salt. In Figure 2 the two modes as seen by the low Q counter bank of the IRS are seen to be clearly separated (the ordinate, S, is proportional to  $S(Q_{i}\omega)$ ); the insert shows the comparable spectrum measured on the beryllium filter spectrometer IN1 at ILL. In addition, in collaboration with Durham Univer-sity (4), the sodium bifluoride was run at a higher incident energy  $E_{O}$  = 450 meV in order to collect data on the second harmonic at  $\sim$  300 meV. In a run of length 78 hours, data were collected over a wide Q range. Figures 3 and 4 show the raw data from the low angle (70) counter and the 24° counter. Under these conditions the resolution is not good enough to separate the bend and stretch modes.

In a collaboration with the Universities of Munster and Birmingham (5) data were taken on samples of vanadium, and vanadium titanium, hydrides. With little multiphonon broadening of the optical mode, data from the high angle counter banks can be summed with little or no loss of resolution. Figures 5 and 6 show reduced data  $(S(Q,\omega))$  against  $\hbar\omega$  for the twosamples with the optical mode clearly split.

#### CONCLUSION

All indications are that the energy resolution is as predicted and is certainly twice as good as that of any other spectrometer presently available to the UK users at these energy transfers. Backgrounds on the high angle banks 240-940 are excellent but at the low angles are too high at present for anything but hydrogenous samples. Tests have shown that much of this background comes from the main beam in the area of the collimation between the chopper and the sample, and steps are being taken to improve this area.

# References

- (1) Boland B C, Mildner D F R, Stirling G C, Bunce L J, Sinclair R N, Windsor C G, Nuclear Instruments and Methods 154 (1978) 349.
- (2) Waddington T C, Howard J, Brierley K P and Tomkinson J, J Chem Phys 64 (1982) 193.
- (3) Salthouse J A and Waddington T C, J Chem Phys 48 (1968) 5274.
- (4) Howard J, Tomkinson J, Boland B C, to be published.
- (5) Severin H, Wilson S K P, Wicke E, Ross D K, Carlile C J, to be published.

## IRS Instrument Parameters

Moderator

5 cm thick water - Cd poisoned

Slab geometry

Viewed at 25° to normal

Area ∿ 170 cm<sup>2</sup> No decoupler

Chopper

At 6.4 metres

12 slot Nimonic operating at 600 Hz

Beam size 2" x 1"

Peak transmission at 300 meV Phased to better than  $\pm \frac{1}{2} \mu s$ 

Sample

At 7.4 metres

In evacuated chamber

Room temperature to 200K Displex type cryostats

Detector

Low angle  $4^{\circ}$ - $11^{\circ}$  either side of main beam

2 x 12 10 atmosphere He<sup>3</sup> counters at 2.5 metres 8 banks 24<sup>o</sup>-94<sup>o</sup> each 2 x 18" 4 atmosphere He<sup>3</sup>

at 1.6 metres

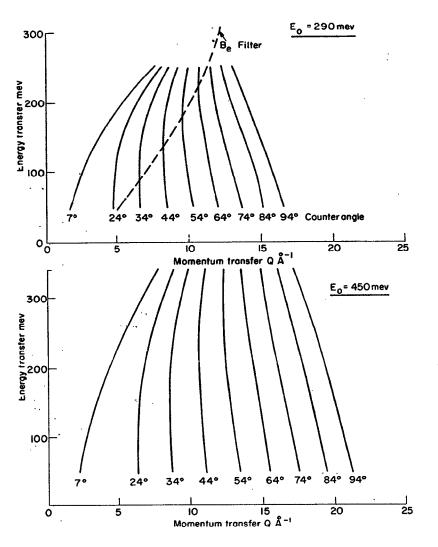
T.O.F.

1024 channels channel width  $l \mu s$ 

variable start delay

Flux on sample

 $\sim 4 \times 10^3$  n/s at full power at E<sub>O</sub> = 450 meV.



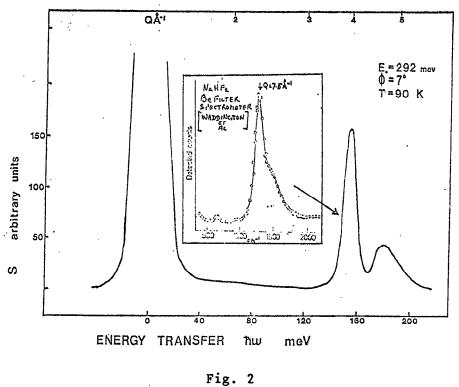


Fig. 1. Inelastic rotor spectrometer.

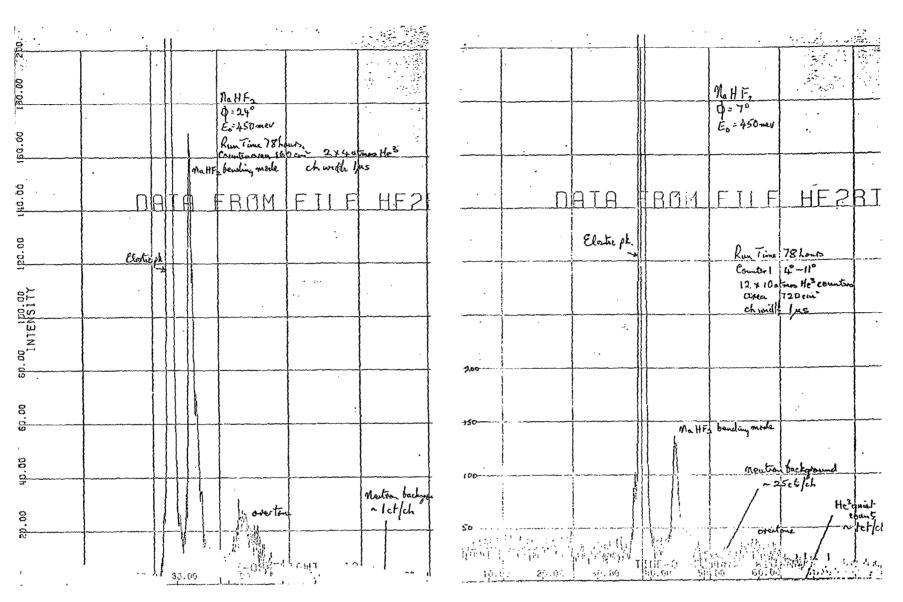
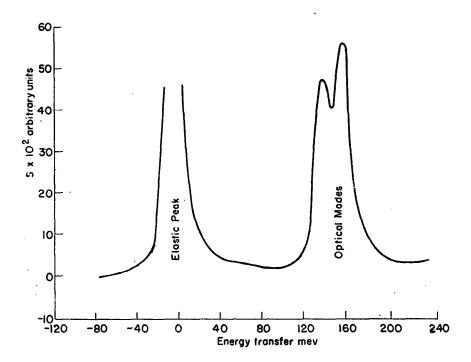


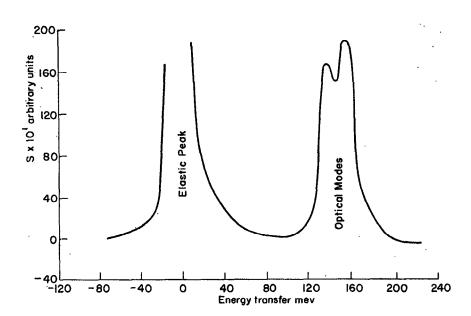
Fig. 3

Fig. 4



Vanadium Hydride
Run time — 48 hours

Fig. 5



Vanadium Titanium Hydride
Run time — 48 hours

Fig. 6

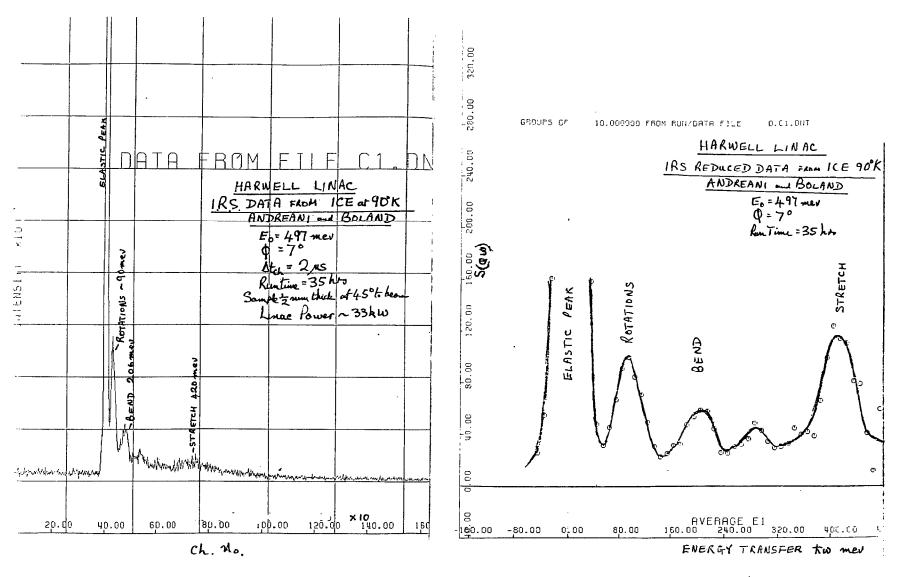


Fig. 7

Fig. 8